

SUPERVISOR'S DECLARATION

I hereby declare that I have checked this report and in my opinion, this report is adequate in terms of scope and quality for the award of the degree of Bachelor of Manufacturing Engineering with honors.



(Supervisor's Signature)

Full Name : DR. RADHIYAH BINTI ABD AZIZ

Position : SENIOR LECTURER

Date : 7 JUNE 2017

DR. RADHIYAH BINTI ABD AZIZ
PENSYARAH KANAN
FAKULTI KEJURUTERAAN PEMBUATAN
UNIVERSITI MALAYSIA PAHANG
26600 PEKAN
PAHANG DARUL MAKMUR
TEL: 09-424 5856 FAKS: 09-424 5888



STUDENT'S DECLARATION

I hereby declare that the work in this report is based on my original work except for quotations and citations which have been duly acknowledged. I also declare that it has not been previously or concurrently submitted for any other degree at Universiti Malaysia Pahang or any other institutions.

A handwritten signature in black ink, appearing to read 'Maisalsadila Binti Ismail', written over a horizontal line.

(Student's Signature)

Full Name : MAISALSADILA BINTI ISMAIL

ID Number : FA13011

Date : 7 JUNE 2017

PERPUSTAKAAN UMP



0000119213

STUDY THE EFFECT OF MARTENSITE-FERRITE CONSTITUENTS PHASE OF
THE PLAIN MEDIUM CARBON DUAL PHASE (DP) STEEL ON THE
MECHANICAL PROPERTY AND CORROSION BEHAVIOR

MAISALSADILA BINTI ISMAIL

Thesis submitted in fulfillment of the requirements
for the award of the degree of
Bachelor of Engineering in Manufacturing Engineering

Faculty of Manufacturing Engineering
UNIVERSITI MALAYSIA PAHANG

JUNE 2017

PERPUSTAKAAN UNIVERSITI MALAYSIA PAHANG	
No. Perolehan 119213	No. Panggilan FCP IM35 2017 r Bc.
Tarikh 09 AUG 2017	

ACKNOWLEDGEMENTS

I am grateful and would like to express my sincere gratitude to my supervisor Dr. Radhiyah Binti Abd Aziz for her germinal ideas, invaluable guidance, continuous encouragement and constant support in making this research possible. She has always impressed me with her outstanding professional conduct, her strong conviction for science, and her belief that a Bachelor program is only a start of a life-long learning experience. I appreciate her consistent support from the first day I'm being introduced as her Final Year Project's student and I am truly grateful for her progressive vision about my training in science, her tolerance of my naive mistakes, and her commitment to my future career.

My sincere thanks go to all my friends and members of the staff of the Manufacturing Engineering Department, UMP, who helped me in many ways and made me stay at UMP pleasant and unforgettable.

I acknowledge my sincere indebtedness and gratitude to my parents and family for their love, dream and sacrifice throughout my life. I cannot find the appropriate words that could properly describe my appreciation for their devotion, support and faith in my ability to attain my goals. Special thanks should be given to my committee members. I would like to acknowledge their comments and suggestions, which was crucial for the successful completion of this study.

ABSTRAK

Kajian dibuat bagi menilai kesan asas fasa ferrite-martensite terhadap keluli karbon sederhana yang melalui proses rawatan haba yang telah diubahsuai terhadap tiga sampel kumpulan dan satu kumpulan sampel yang tidak dirawat di dalam asid hidroklorik 1M. Setiap kumpulan sampel tersebut telah dilabel sebagai kumpulan A (bagi sampel yang tidak dirawat untuk dijadikan sebagai rujukan), B, C and D. Pada permulaan proses, Keluli karbon sederhana kumpulan C dan D akan dipanaskan sehingga suhu 950 °C supaya mencapai paras austenite yang sebenarnya (γ -fasa) dan kemudian disejukkan di dalam air manakala sampel kumpulan B akan dipanaskan sehingga suhu 750 °C bagi menghasilkan dua fasa ferrite-martensite dan seterusnya direndam di dalam air bagi proses penyejukan. Kemudian, keluli kumpulan C dan D akan dipanaskan kembali sehingga suhu 750 °C seperti keluli kumpulan B iaitu bagi menghasilkan dua fasa ferrite-martensite dan seterusnya keluli tersebut direndam di dalam air bagi proses penyejukan. Keluli karbon tersebut kemudian dipanaskan sekali lagi pada suhu yang lebih rendah iaitu 480 °C. Ciri-ciri hakisan yang berlaku pada keluli karbon yang direndam di dalam 1M asid hidroklorik (HCL) telah dikaji dengan mengukur kadar penurunan berat setiap sampel tersebut. Bagi Keluli karbon sederhana yang tidak dirawat, kehilangan berat adalah antara 0.003g – 0.022g manakala keluli karbon sederhana haba yang dirawat pula kehilangan berat adalah antara 0.004g – 0.167 g. Ujian mekanikal telah dijalankan ke atas sampel kumpulan A, B, C dan D. Keputusan yang diperolehi menunjukkan bahawa kekerasan sampel itu bergantung pada kadar jumlah martensite yang hadir di dalam setiap sampel yang mempunyai dua fasa iaitu ferrite-martensite.

ABSTRACT

The investigation was carried out to study the effect of martensite-ferrite constituent phase on the medium carbon steels that undergo modified heat treatment process for three groups of samples and a group of untreated sample on corrosion test 1M hydrochloric acid. Each of the group samples were labeling with group A (untreated sample act as reference), B, C and D. The medium carbon steels for samples group C and D were heated to a temperature of 950 °C for completely austenite (γ -phase) while samples group B were heated until temperature 750 °C to form ferrite-martensite dual phase region and then quenched in water. Then, group C and D steels were reheated until temperature 750 °C of ferrite-martensite dual phase region and rapidly quenched in water. Next, the steels were tempered at a temperature of 480 °C. The corrosion behavior of the carbon steel in 1M hydrochloric acid (HCL) was studied by measuring the weight loss of the steels. For untreated medium carbon steel, the weight loss is between 0.003g – 0.022g while for heat treated medium carbon steel, the weight loss is between 0.004g – 0.167g. The mechanical test was carried out on the samples of group A, B, C and D. The results obtained indicate that the specimen hardness is proportional to the amount of martensite in dual phase steel.

TABLE OF CONTENT

DECLARATION	
TITLE PAGE	
ACKNOWLEDGEMENTS	ii
ABSTRAK	iii
ABSTRACT	iv
TABLE OF CONTENT	v
LIST OF TABLES	viii
LIST OF FIGURES	ix
LIST OF SYMBOLS	x
LIST OF ABBREVIATIONS	xi
CHAPTER 1 INTRODUCTION	1
1.1 Background	1
1.2 Problem Statement	2
1.3 Objectives of Research	3
1.4 Scope of Research	3
1.5 Statement of Contribution	4
CHAPTER 2 LITERATURE REVIEW	5
2.1 Introduction	5
2.2 Ferrous Metal Alloy	5
2.2.1 Carbon Steels	5
2.2.2 Iron-iron Carbide (Fe-Fe ₃ C)	7

2.3	Corrosion Behaviour of Steel	9
2.3.1	Prevention of Corrosion	9
2.4	Heat Treatment	10
2.4.1	Annealing	11
2.4.2	Normalizing	11
2.4.3	Quenching	11
2.4.4	Tempering	11
2.5	Dual-Phase Steels	12
CHAPTER 3 METHODOLOGY		14
3.1	Introduction	14
3.2	Research Methodology	14
3.3	Machine	16
3.3.1	Sectioning Cut-Off Machine	16
3.3.2	Nabertherm furnace Machine	17
3.4	Location of Experiment	18
3.5	Process and Procedure involve	18
3.5.1	Preparation of specimen medium carbon steels	18
3.5.2	Heat Treatment Modification Process	19
3.5.3	Microstructure Examination	22
3.5.4	Mechanical Testing	24
3.5.5	Corrosion Test Evaluation	25
3.5.6	Corrosion Rate (CR)	27
CHAPTER 4 RESULTS AND DISCUSSION		28
4.1	Introduction	28

4.2	Microstructural Examination	28
4.3	Mechanical Property	29
4.4	Corrosion Test	30
CHAPTER 5 CONCLUSIONS & RECOMMENDATIONS		33
5.1	Introduction	33
5.2	Conclusions	33
5.3	Recommendations for the Future Research	34
REFERENCES		35
APPENDIX A GANTT CHART OF FINAL YEAR PROJECT		38
APPENDIX B RAW DATA FROM THE EXPERIMENT		40

LIST OF TABLES

Table 2	Recent research on corrosion prevention for carbon steel	9
Table 4.1	Average result of hardness value for the samples	30
Table 4.2	Weight loss and corrosion rate of sample A, B, C and D for Medium carbon steel	31

LIST OF FIGURES

Figure 2.1	An illustration of Iron-Iron Carbide (Fe-Fe ₃ C) Phase Diagram	7
Figure 2.2	An illustration of Iron-iron carbide (Fe-Fe ₃ C) phase diagram	12
Figure 3.1	Process flow of prepared and conducted experiment	15
Figure 3.3	Abrasive cut-off blades	16
Figure 3.4	Heat treatment processes by using furnace	17
Figure 3.5	Glove and crucible tongs	17
Figure 3.7	The material of medium carbon steel after cut into smaller size	19
Figure 3.8	Temperature – Time Graph involving Lamellarizing and Tempering (L+T)	20
Figure 3.9	Temperature – Time Graph involving Quenching, Lamellarizing and Tempering (Q+L+T)	21
Figure 3.10	Temperature – Time Graph involving Quenching, Quenching, Lamellarizing and Tempering (Q+Q+L+T)	22
Figure 3.11	Grinding process of specimens using Metkon Grinder- Polisher machine	22
Figure 3.12	Different grades of emery papers	23
Figure 3.13	Polishing process using Topper Grinding-Polishing Machine	24
Figure 3.14	Check microstructure of specimen using optical microscope and Optic Images Plus 2.0 software	24
Figure 3.15	Digital Metallic Vickers Hardness Tester	25
Figure 3.16	1 M dilute hydrochloric acid solution	26
Figure 3.17	Electronic weighing machine	27
Figure 4.1(a) – (d):	Revealed microstructures using 50X magnification of optical microscope of the phases that are present under different group of heat treatment process.	29
Figure 4.2	Average Vickers hardness of un-treated and heat treatment samples	30
Figure 4.3	Relationship between weight loss and duration of medium carbon steel immersed in HCl solution for sample A, B, C and D	31
Figure 4.4	Relationship between corrosion rate and duration of medium carbon steel immersed in HCl solution for sample A, B, C and D	32

LIST OF SYMBOLS

$\%$	Percentage
α	Alpha
γ	Gamma
δ	Delta
$^{\circ}\text{C}$	Degree Celsius
$^{\circ}\text{F}$	Degree Fahrenheit

LIST OF ABBREVIATIONS

UTS	Ultimate Tensile Strength
DP	Dual-phase

CHAPTER 1

INTRODUCTION

1.1 Background

The most useful and highly applicable engineering material is a carbon steel which represented around 85% of the aggregate yearly steel generation around the world (Tolulope, Akintoye, Joseph, & Olanrewaju, 2016). It is the major classification of metallic alloys which applied in tonnage and total cost. Corrosion of carbon steel has been an issue of enormous practical significance because of its high cost on the national economy. In spite of its moderately constrained corrosion resistance, carbon steel is utilized as a part on marine applications, pipelines, mining, chemical processing, metal-processing equipment, production and refining of petroleum, chemical processing, pipelines, mining, construction, fossil fuel power and nuclear power plants. However, due to the thermodynamic instability of carbon steels, they are strongly susceptible to corrosion when exposed to aqueous environments. This is more apparent in acidic media as more applications of acid solution in most industrial field. The corrosion issue consists of the major portion of the overall expenditure for petrochemical companies worldwide, occurring whenever stages from down-hole to surface equipment and processing facilities. There are serious economic losses when the corrosion induced as it is usually associated with operational problems and needs maintenance of plants and equipment continuously but under limited or complete process shut- down (Tolulope et al., 2016).

Appropriate corrosion control and prevention methods have been shown to help mitigate against potential disasters that capable causes negative social impacts, environmental pollution and damage to water resource. This is one of the reasons that the iron and steel get a protective system from the existence of entire industries. Historically, corrosion inhibitors have been observed to have excellent anti-corrosive

properties; however, a significant proportion of them caused secondary effect, damaging the environment. In order to protect the metals from corrosion in atmospheric conditions, one of the prevention methods that are widely used is organic paint coatings. Thus, as the demand for thermal-resistant coating increases, but there are some questionable about their thermal stability at high-temperature (Nazeri, Suan, Masri, & Alias, 2012).

Generally, the composition of the alloy, heat treatment and mechanical processing are involving much in phase transformation behaviour, mechanical properties, corrosion resistance and processing (Renato et al., 2014). Hence, the corrosion rate of carbon steel is also can be influenced by its chemical composition and microstructure, so that it's not only governed by the electrolyte conditions. Therefore, to correlate the metallurgical concept with the corrosion parameters, the corrosion behaviour of carbon steel and effects of microstructure on such behaviour is still an open field for investigation. Heat treatment condition would have a great effect on the microstructure of the carbon steel and directly connected to the corrosion resistance of the carbon steel. Thus, the purpose of this study is to establish heat treatment conditions that can give optimum microstructure phases which contribute to the least corrosion rate of the carbon steel.

1.2 Problem Statement

Dual-phase (DP) steels are being used in automobile industries for last three decades. DP steels possessing a composite microstructure consisting of hard martensite islands embedded in a soft ferritic matrix have evoked much interest. DP steels possess a number of unique properties such as continuous yielding, low 0.2% offset yield strength, high ratio of ultimate tensile strength (UTS) to yield stress, high work hardening rate, and high uniform and total elongations, which are making them attractive for applications such as very good quality sheet materials for automotive bodies. However, in order to meet other requirements such as durability of sheet metal products over long periods, the corrosion resistance becomes crucial. There is research has studied electrochemical behaviour of micro alloyed DP steels and found that with increase in martensite content and structural refinement, the corrosion resistance decreased. However, there is little work on corrosion behaviour of plain medium carbon DP steels. Further investigations in this direction are necessary to find the effect of

martensite volume fraction of plain medium DP steel on its corrosion behaviour. The research investigation deals with the study of mechanical properties (hardness) and corrosion behaviour of medium carbon DP steels with different fractions of martensite and ferrite phase. The constituent phase is controlled by varying the heat treatment process. These results are also compared with the corrosion behaviour of as received steel with ferrite and pearlite as micro constituents.

1.3 Objectives of Research

The main objectives of this project are:

- i. To study the effect of heat treatment process on the microstructure of medium carbon steel materials.
- ii. To evaluate the effect of microstructure changes on the mechanical property.
- iii. To investigate the effect of martensite-ferrite phase on the corrosion behaviour of medium carbon steel.

1.4 Scope of Research

In order to achieve the objectives, the following scopes of studies are performed:

- i. The study will conduct annealing, quenching, lamellarizing and tempering of heat treatment process.
- ii. The effect of heat treatment on corrosion rate will be varied by testing on 1M hydrochloric acid (HCl) solution.
- iii. Mechanical testing including hardness test, corrosion test and microscopic examination will be conducted in this study.

REFERENCES

This thesis is prepared based on the following references;

- Ayodele, C. O., & Nenuwa, O. B. (2013). Investigation of the Effects of Heat Treatment on the Corrosion Behaviour of Welded Low Carbon Steel in Different Environments, *3*(8), 333–340.
- Bai, X., Tang, J., Gong, J., & Lv, X. (2016). Corrosion Performance of Al-Al₂O₃ Cold Sprayed Coatings on Mild Carbon Steel Pipe under Thermal Insulation. *Chinese Journal of Chemical Engineering*.
- Bhagavathi, L. R., Chaudhari, G. P., & Nath, S. K. (2011). Mechanical and corrosion behavior of plain low carbon dual-phase steels. *Materials and Design*, *32*(1), 433–440.
- Callister, W., & Rethwisch, D. (2007). *Materials science and engineering: an introduction. Materials Science and Engineering* (Vol. 94).
- Durmaz, H., Beril Tugrul, A., & Buyuk, B. (2014). Evaluation of gamma penetration through plain carbon steels. *Acta Physica Polonica A*, *125*(2), 469–472.
- El Azzouzi, M., Aouniti, A., Tighadouin, S., Elmsellem, H., Radi, S., Hammouti, B., ... Zarrouk, A. (2016). Some hydrazine derivatives as corrosion inhibitors for mild steel in 1.0 M HCl: Weight loss, electrochemical, SEM and theoretical studies. *Journal of Molecular Liquids*, *221*, 633–641.
- Farag, A. A., Migahed, M. A., & Al-Sabagh, A. M. (2015). Adsorption and inhibition behavior of a novel Schiff base on carbon steel corrosion in acid media. *Egyptian Journal of Petroleum*, *24*(3), 307–315.
- Hamdy, A., & El-Gendy, N. S. (2013). Thermodynamic, adsorption and

- electrochemical studies for corrosion inhibition of carbon steel by henna extract in acid medium. *Egyptian Journal of Petroleum*, 22(1), 17–25.
- Hartwig, A., Decker, M., Klein, O., & Karl, H. (2015). Stoichiometric titanium dioxide ion implantation in AISI 304 stainless steel for corrosion protection. *Nuclear Instruments and Methods in Physics Research, Section B: Beam Interactions with Materials and Atoms*, 365, 94–99.
- Jha, B. K., Avtar, R., & Sagar Dwivedi, V. (1996). Structure-property correlation in low carbon low alloy high strength wire rods/wire containing retained austenite. *Transactions of the Indian Institute of Metals*, 49(3), 133–142.
- Kim, S. T., Jang, S. H., Lee, I. S., & Park, Y. S. (2011). Effects of solution heat-treatment and nitrogen in shielding gas on the resistance to pitting corrosion of hyper duplex stainless steel welds. *Corrosion Science*, 53(5), 1939–1947.
- Messien, P., Herman, J.-C., & Greday, T. (1981). Phase Transformation and microstructures of intercritically annealed Dual-Phase Steels. *Fundamentals of Dual-Phase Steels*, 161–180.
- Nadlene, R., Esah, H., Norliana, S., & Irwan, M. A. M. (2011). Study on the Effect of Volume Fraction of Dual Phase Steel to Corrosion Behaviour and Hardness, 5(2), 393–396.
- Nazeera Banu, V. R., Rajendran, S., & Senthil Kumaran, S. (2016). Investigation of the inhibitive effect of Tween 20 self assembling nanofilms on corrosion of carbon steel. *Journal of Alloys and Compounds*, 675, 139–148.
- Ndaliman, M. B. (2006). An Assessment of Mechanical Properties of Medium Carbon Steel under Different Quenching Media. *J.T.*, 10(2)(2), 100–104.
- Sarkar, P. P., Kumar, P., Manna, M. K., & Chakraborti, P. C. (2005). Microstructural

- influence on the electrochemical corrosion behaviour of dual-phase steels in 3.5% NaCl solution. *Materials Letters*, 59(19–20), 2488–2491.
- Shalash, L., & Nasher, L. (2010). Study the effect of magnetic field on the corrosion of steel in sodium chloride solution (NaCl). *Researchgate.Net*, 9(17), 30–38.
- Speich, G. R., Demarest, V. A., & Miller, R. L. (1981). Formation of Austenite During Intercritical Annealing of Dual-Phase Steels. *Metallurgical Transactions A*, 12(8), 1419–1428.
- Thomas, G. (1996). Dual Phase Steel Revisited: Reinforcement for Concrete. *Transactions of the Indian Institute of Metals*, 49(3), 127–132.
- Tolulope, R., Akintoye, C., Joseph, O., & Olanrewaju, G. (2016). Adsorption and corrosion inhibition properties of thiocarbanilide on the electrochemical behavior of high carbon steel in dilute acid solutions. *Results in Physics*, 6, 305–314.
- Toro, A., Tanaka, D., Sinatora, A., & Tschiptschin, A. (2001). Improvement of Corrosion-Erosion Resistance of Martensitic Stainless Steels by Nitrogen Addition at High Temperature. *Journal of Materials Processing Technology*, 117(3).
- Yanagisawa, K., Nakanishi, T., Hasegawa, Y., & Fushimi, K. (2015). Passivity of Dual-Phase Carbon Steel with Ferrite and Martensite Phases in pH 8.4 Boric Acid-Borate Buffer Solution. *Journal of the Electrochemical Society*, 162(7), C322–C326.
- Zheng, Z. B., & Zheng, Y. G. (2016). Effects of surface treatments on the corrosion and erosion-corrosion of 304 stainless steel in 3.5% NaCl solution. *Corrosion Science*, 112, 657–668.